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Olefinic stereoselection in the [2,3]-Wittig rearrangement of α,β -disubstituted allylic ethers forming trisubstituted olefins

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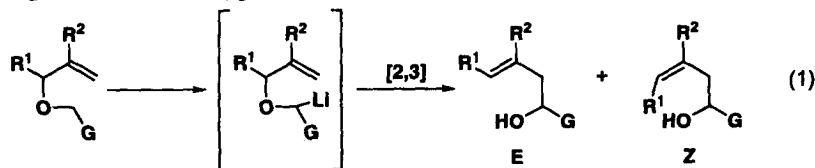
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Abstract

The *E/Z*-selectivities in the [2,3]-Wittig rearrangements of secondary β -(methyl or silyl)allylic ethers are shown to depend critically on the nature of groups on the carbanion terminus, thereby permitting elucidation of the structural requirements for attaining high *Z*-selectivity. © 1999 Elsevier Science Ltd. All rights reserved.

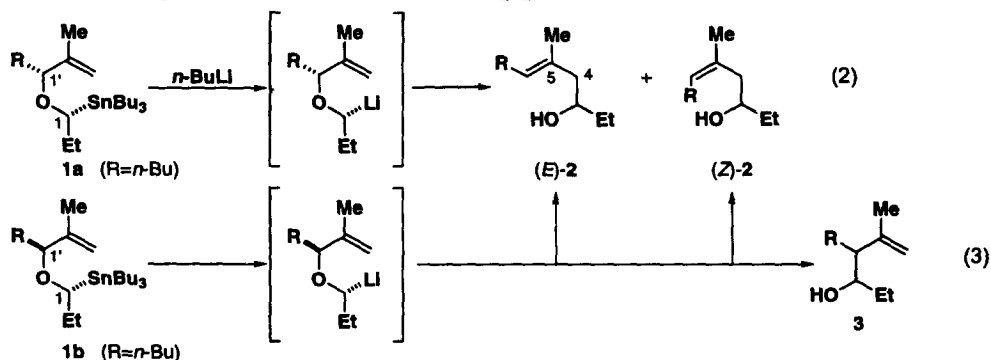
Keywords: [2,3]-Wittig rearrangement; olefinic stereoselection; *Z*-trisubstituted olefins.

While the [2,3]-Wittig rearrangement of secondary allylic ethers generally exhibits a high *E*-selectivity over the newly created olefin bond, several exceptions to the *E*-selection attribute have been reported, in particular, with respect to α,β -disubstituted allylic ethers (Eq. 1).¹ The most notable is the Wittig–Still variant ($G=H$, R^1 =alkyl, R^2 =Me) which shows a remarkably high level (>95%) of *Z*-selection² and hence has found applications for *Z*-trisubstituted olefin synthesis.¹ Another notable exception is the rearrangement of the β -(silyl)allylic propargyl ether ($G=C\equiv C\text{SiMe}_3$, R^1 =pentyl, R^2 =SiMe₃) which provides 80% '*Z*'-selectivity,^{3,4} while the allyl counterpart ($G=CH=CH_2$, R^1 =methyl or pentyl) shows only 36–33% '*Z*'-selectivity.^{4,5} While these high *Z*-selectivities have been interpreted as a result of the alleviation of the steric 1,2-repulsion between R^1 and R^2 in the transition states, these examples point out that the nature of *G* group must be considered as another key factor in dictating the level of *Z*-selectivity.⁶ Thus, the question arises as to how the nature of the *G* group affects the *E/Z*-selectivity in the [2,3]-Wittig rearrangement in general or what the requisite structural factors are for attaining high *Z*-selectivity. Herein we wish to address this fundamental question based on the *E/Z*-selections observed in the [2,3]-rearrangements of two types of α,β -disubstituted allylic ethers with different *G* groups.

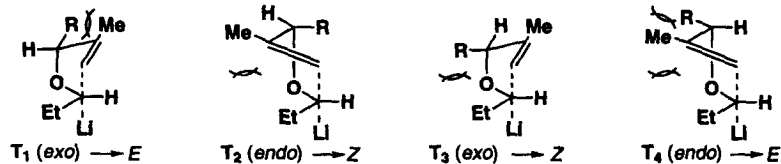


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First, we examined the *E/Z*-selectivity in the Still-type rearrangement of the diastereomeric pair of the methallylic ether **1a** and **1b** where the transmetalation proceeds with complete retention of configuration and the Li-bearing terminus is configurationally stable⁷ (Eqs. 2 and 3). Diastereo-defined substrates **1a** and **1b** were obtained by column-chromatographic separation and the relative configuration was assigned by ¹H NMR comparison with an authentic (1*R*, 1'*S*)-enantiomer of **1b** prepared via the reaction of the mesylate of (*R*)- α -hydroxypropylstannane with the potassium salt of (*R*)-2-methyl-2-hepten-3-ol.^{7,8} Thus, **1a** was treated with *n*-BuLi in THF at -78°C to give the (*E*)-olefin **2** as a single isomer in 76% yield, whereas a similar rearrangement of **1b** was much slower to give rise to an isomeric mixture (*E*:*Z*=24:76) of **2** in 25% yield, along with 16% of the [1,2]-Wittig product **3**.⁹

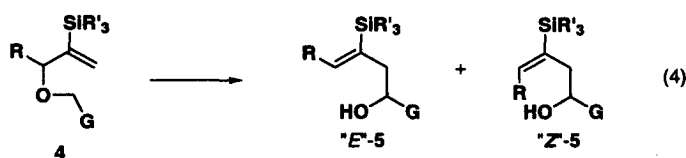


These outcomes reveal that the introduction of ethyl as the G group to the original Still system (G=H) no longer affords high *Z*-selectivity. This rather surprising observation suggests that a steric factor offered by the Et group predominates over the steric 1,2-repulsion between Me and R. Between the two transition states **T**₁(*exo*) and **T**₂(*endo*) available for **1a**,¹⁰ the former, despite the presence of the 1,2-repulsion, might be sterically much more favorable because the latter suffers a 1,3-repulsion between Et and Me. For the other diastereomer **1b**, on the other hand, while both **T**₃(*exo*) and **T**₄(*endo*) are sterically disfavored, the former being free of the 1,2-repulsion is more favorable, thus leading to the *Z*-selection. Thus, it is safe to say that the present rearrangement proceeds preferentially through the *exo*-transition states, either in the presence or absence of the 1,2-repulsion. That means that the *exo*-preference possessed inherently by the Et group prevails as a stereo-directing factor over the 1,2-repulsion. In other words, if the *exo/endo* complication is absent, the 1,2-steric repulsion would become the sole stereo-directing factor, thus leading to high *Z*-selectivity. That is exactly the case of the original Still variant (G=H).



Next, our attention was focused on *E/Z*-selection in the rearrangements of secondary β -(silyl)allylic ethers **4** where the Li-bearing terminus is configurationally labile (Eq. 4). The '*E/Z*'-selectivities thus observed are summarized in Table 1, along with the literature data for comparison. The most revealing is that the use of silylethynyl as the G group provides a significantly higher *Z*-selectivity than those of the vinylic ones (entry 2 vs. 8). Rather interestingly, the bulkiness of the β -silyl groups has little effect (entries 5–7). A remarkably high '*Z*'-selectivity was attained when G was a silylethynyl and R was a bulky alkyl (entries 8 and 9). These trends are explicable by considering that the silylethynyl group is known as one of the few G groups that possess *endo*-preference in the transition states, whereas the vinylic

Table 1
The [2,3]-Wittig rearrangement of β -(silyl)allylic ethers^a



Entry	Substrate 4			Base	"E"/"Z" ^b	%yield
1 ^c	G=CH=CH ₂ ,	R=Me,	R'=Me	LDCA ^d	64 : 36	62
2 ^e		R= <i>n</i> -C ₅ H ₁₁		<i>n</i> -BuLi	67 : 33	78
3	G=C(Me)=CH ₂ ,	R=Me,	R'=Me	LDCA ^d	59 : 41	87
4 ^c				<i>n</i> -BuLi	95 : 5	35
5	G=C≡CSiMe ₃ ,	R= Me,	R'=Me	<i>n</i> -BuLi	46 : 54	98
6			R'=Et		44 : 56	99
7			R'=Ph		52 : 48	98
8	G=C≡CSiMe ₃ ,	R= <i>n</i> -C ₅ H ₁₁ ,	R'=Me	<i>n</i> -BuLi	22 : 78	98
					(20 : 80) ^f	
9	G=C≡CSiMe ₃ ,	R= <i>i</i> -Pr,	R'=Me	<i>n</i> -BuLi	1 : 99 ^f	87
10 ^g	G=H (SuBu ₃),	R=Me,	R'=Me	<i>n</i> -BuLi	1 : 99 ^f	99

^a The reactions were carried out in THF at -78 °C. ^b The geometry was assigned by ¹H NMR spectra on the basis of the empirical rule for the olefinic protons' peaks (refs. 5 and 11), and the ratio was determined by ¹H NMR and/or GLC. ^c Cited from ref. 5. ^d Lithium dicyclohexylamide. ^e Cited from ref. 3. ^f The "Z" geometry of the major product was confirmed by its conversion to the corresponding Z-disubstituted olefin via protodesilylation (refs. 5 and 11b). ^g Performed using Still's transmetalation method.

groups have *exo*-preference.^{1,12} Thus, it appears likely that the relatively high *Z*-selectivity observed with G=C≡CSiR₃ reflects the special situation where the *endo*-TS corresponding to T₂ (free of 1,2-repulsion) prevails overwhelmingly over the other *endo*-TS corresponding to T₄ (suffering 1,2-repulsion). Thus, it is safe to conclude that a high *Z*-selectivity can be attained only when one employs G=H (where the *endo/exo* complication is absent) or such G group as silylethynyl that possesses a large *endo*-preference in the transition states. In other words, the *E/Z*-selectivity of the [2,3]-Wittig rearrangement in general is determined by the balance between the magnitude of the 1,2-repulsion between R¹ and R² and the *exo/endo*-preference of G-group used. The more *endo*-preference, the more *Z*-selectivity.

In summary, we have demonstrated that the *E/Z*-selectivity in the [2,3]-Wittig rearrangements of α,β -disubstituted allylic ethers depends critically upon the nature of G groups. Moreover, we have clarified not only the mechanistic origin of the high *Z*-selectivity reported for the Wittig–Still variant (G=H) and G=C≡CSiMe₃, but also the requirements for attaining high *Z*-selectivity. Further work is in progress to develop other *Z*-selective [2,3]-Wittig variants.

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4. Throughout this paper, the descriptor 'Z' is used for (*E*)-silylolefin products (where R¹ and R²=SiR₃ are *trans*) for the sake of consistency. Thus, note that 'E' corresponds to Z (by the usual convention).
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8. The two substrates are unequivocally distinguishable by ¹H NMR spectra: e.g., δ (ppm, CDCl₃) for 1-H, 3.69 (dd, J=5.8, 7.4 Hz) for **1a** and 3.47 (dd, J=6.4, 7.4 Hz) for **1b**.
9. The product geometry was assigned by ¹³C NMR spectra on the basis of the 'γ-effect' on allylic carbons (cf. Ref. 2). The most distinguishable are the peaks (ppm) due to 5-Me and C4: 16.1 and 47.6 for (*E*)-**2** and 23.8 and 39.7 for (*Z*)-**2**. Byproduct **3**: ¹H NMR, δ 0.88 (t, J=7.2 Hz, 3H), 0.98 (t, 7.4 Hz, 3H), 1.10–1.42 (m, 8H), 1.59 (br. s, 1H), 1.68 (s, 3H), 2.04 (ddd, J=4.4, 8.5, 10.5 Hz, 1H), 3.34–3.42 (m, 1H), 4.81–4.84 (m, 1H), 4.95–4.98 (m, 1H).
10. Note that the [2,3]-Wittig rearrangement is well-established to proceed with complete inversion of configuration at the Li-bearing terminus (Ref. 7a).
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